"APPROVED FOR RELEASE: Thursday, September 26, 2002 APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3 CIA-RDP86-00513R000515110005-3"

GIL'MAN, T.P.; PERLIN, S.M.; LETTES, A.Z.

Electro consistemeter for determining the processing time, gelatinisation, and hardening of resins. Plast.massy no.11:68-71 '60. (MIRA 13:12)

(Resins, Synthetic)

S/069/62/024/002/004/008 B101/B110

15.8310

AUTHORS: Zubov, P. I., Lepilkina, L. A., Gil'man, T. P.

TITLE:

Effect of lubricant and finishing materials on the internal stresses and adhesion properties of polyester coatings

PERIODICAL:

Kolloidny shurnal, v. 24, no. 2, 1962, 174-177

TEXT: $\Pi H-1$ (PN-1) polyester resin films, $\sim 2200~\mu$ thick, were applied to glass parallelepipeds and polymerized at $75^{\circ}C$ in the presence of 3% cumene hydroperoxide and 8, cobalt naphthenate dissolved in styrene. One of the glass surfaces was modified with a preparation, and the internal stress was measured optically with a self-recording instrument. Adhesion was determined from the maximum (critical) stress at which the film detached from the glass. The following modifiers were used: (1) Paraffin emulsion consisting of stearin, vaseline, and transformer oil with CO-20 (SO-20) dicyana diamine formaldehyde resin as emulgator: the film detached already after 30 min. (2)AC-1 (AS-1) disapol, a polymerization product from butyl methacrylate and methacrylamide in the presence of dibutyl sebacinate: here, and on unmodified surfaces, at lower internal stress, however, separation set in after 12 hrs. (3)M Φ -17 (MF-17) urea formaldehyde resins Card 1/3

s/069/62/024/002/004/008

Effect of lubricant and ...

showed better results: film adhesion to glass exceeded 12 hrs. (4) The best results were obtained with MB3-3 (PVE-3) polyvinyl acetate emulsion with and without chromolan additions (a cation-active preparation). Internal stress increased after 30-60 min but was moderated by 0.7% chromolan. Then, gradual relaxation followed. The film did not detach from the glass. Tests for the effect of film thickness on its separation from the glass yielded similar results from the different preparations: from glass modified with paraffin emulsion, a film thinner than that from unmodified glass detached, whereas with MF-17 thicker films showed good adhesion. Data are given for glass reinforced plastics with a 50% content of glass fiber: the bending strength (a) and internal stress (b) obtained with paraffin emulsion were 2200 kg/cm² and 10.8 kg/cm², respectively; with MF-17 a = 2880, b = 28.6; with AS-1 a = 2596, b = 3.8, and with PVE-3 containing 0.7% chromolan, a = 3300, b = 2.8. There are 4 figures, 1 table, and 2 Soviet references.

ASSOCIATION: Institut fizicheskoy khimii AN SSSR, Otdel polimerov (Institute of Physical Chemistry of AS USSR, Department of Polymers). Vsesoyuznyy nauchno-issledovatel'skiy proyektnyy institut ugol'nogo mashinostroyeniya, Moskva (All-Union Scientific Research, Design and Planning Institute of Coal,

Card 2/3

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3" CIA-RDP86-00513R000515110005-3"

Effect of lubricant and ...

S/069/62/024/002/004/008 B101/B110

SUBMITTED:

April 20, 1961

Card 3/3

8/653/61/000/000/034/051 1007/1207

AUTHORS:

Perlin, S.M., Gil'man, T.P., and Leytes, A.Z.

TITLE:

Determination of hardening degree of unsaturated

polyester resins by the dilatometric method

SOURCE:

Plastmassy v mashinostroyenii i priborostroyenii. Pervaya resp. nauch.-tekh. konfer. po vopr. prim. plastmass v mashinostr. i priborostr., Kiev, 1959. Kiev, Gostelhizdat, 1961, 367-375

TEXT: The paper presents results of dilatometric determinations of series of physicomechanical properties of polyester resins by means of the differential dilatometer of the Chevenard system which yields much better results than conventional dilatometers. As was found, hardness, water-absorption and bending strength depend on the hardening degree of the resin. The dilatometric method permits suit-

Card 1/2

S/653/61/000/000/034/051 I007/I207

Determination of hardening degree ...

able evaluation of the hardening degree of the above resins; it makes it also possible to distinguish between the temporary incomplete hardening and the constant incomplete hardening. The above method may also be successfully used for the determination of the hardening degree of glass-reinforced plastics, of their dimensional stability and heat resistance. There are 7 figures.

Card 2/2

"APPROVED FOR RELEASE: Thursday, September 26, 2002
APPROVED FOR RELEASE: Thursday, September 26, 2002
KLINOV, I.Ya.; KUTSKNOK, B.I.; FABRIKANT, T.L., GIL, MAN, TS.I.

Chemically stable mastics based on a modified asbestos vinyl.

Plast.massy no.2:44-50 '61. (MIRA 14:2)

(Plastics) (Protective coatings)

18 8300 /4016, 1138, 1208

S/020/61/137/003/025/030 B101/B208

AUTHORS:

Kolotyrkin, Ya. M., and Gil'man, V. A.

TITLE:

Effect of chlorine ions on the electrochemical and

corrosion behavior of zirconium

PERIODICAL:

Doklady Akademii nauk SSSR, v. 137, no. 3, 1961, 642-645

TEXT: It was found in papers by E. A. Gee, L. B. Golden, W. E. Lusby (Ref. 1, see below), D. F. Taylor (Ref. 2, see below), and L. B. Golden, I. R. Lane, W. L. Acherman (Ref. 3, see below) that zirconium may be corroded by chlorine ions under certain conditions. As these papers do not permit exact conclusions on the causes of this behavior of zirconium, a more thorough investigation has now been made of the conditions, under which Zr is corroded by chlorine ions. The dependence of the dissolution rate of Zr on the potential was determined by a potentiostatic method described by the authors in Ref. 7 (ZhFKh, 30, 1990, (1956)) and Ref. 8 (DAN, 114, 1265 (1957)). The experiments were performed in 1.0; 0.1; 0.01 N HCl; 1.0 N H₂SO₄, 1.0 N KBr; 1.0 N KI. Pure zirconium (99.8%)

Card 1/5

Effect of chlorine ions on the ...

S/020/61/137/003/025/030 B101/B208

was used as electrode. The liquid reagents were purified by distillation. The solutions were saturated with No which was bubbled through also during the measurement. Fig. 1 shows the result of the potentiostatic measurements. In H2SO4, Zr was passive in the entire potential range studied. In the presence of halogen compounds, however, Zr is dissolved when a critical potential $\varphi_{\rm cr}$ is attained, $\varphi_{\rm cr}$ remaining constant irrespective of current density. The following results were obtained in galvanostatic measurements: Temporary positive and negative shifts of the potential are accomplished by increasing and reducing the current density, respectively. The potential always returns to the value ϕ_{cr} Measurement of the charge curves also indicated that at first Zr is polarized to more positive values than $arphi_{ ext{cr}}$. At a constant concentration of Cl, the deviation of the potential from or increases with the current density. At constant current density, the deviation increases with decreasing Cl concentration. Addition of Fe $^{3+}$ exerted the same effect as application of anodic polarization. ϕ_{cr} was attained at a certain Card 2/5

Effect of chlorine ions on the ...

S/020/61/137/003/025/030 B101/B208

concentration of the iron salt. Further increase of the concentration of Fe $^{3+}$ had no influence. It was found visually that, when $\varphi_{\rm cr}$ is attained, an irregular corrosion occurs, which gives rise to the formation of pittings, which increases with the current density. With decreasing current density, the pittings are again partly passivated. This reversibility of the process is explained by the fact that at a certain density of the polarization current, the affinity of Zr to the halogen ion is greater than to the passivating oxygen. The passivating oxygen is displaced by the halogen ion, The irregular corrosion may be explained by the permanent nonuniform distribution of the plate current on the metal surface. The assumption that the corrosion process is retarded in time by the formation of primary complexes of the $ZrCl_n^{(4-n)+}$ type could not be experimentally confirmed. It may therefore be assumed that these complexes decompose by hydrolysis, the chlorine ions are again liberated, and thus act as catalysts of corrosion. Mention is made of N. A. Balashova and B. N. Kabanov (Ref. 15: DAN, 121, 126 (1958)) and L. V. Vanyukova (Ref. 14: DAN 59, 917 (1948)).

Card 3/5

Effect of chlorine ions on the ...

S/020/61/137/003/025/030 B101/B208

There are 3 figures and 15 references: 8 Soviet-bloc and 7 non-Soviet-bloc. The 4 references to English-language publications read as follows: E. A. Gee, L. B. Golden, W. E. Lusby, Ind. and Eng. Chem. 41, 1668 (1949); D. F. Taylor, Ind. and Eng. Chem. 42, 639 (1950); L. B. Golden, I. R. Lane, W. L. Acherman, Ind. and Eng. Chem. 44, 1930 (1952); 45, 782 (1953), I. R. Lane, L. B. Golden, W. L. Acherman, Ind. and Eng. Chem. 45, 1067, (1953).

ASSOCIATION: Fiziko-khimicheskiy institut im. L. Ya. Karpova

(Physicochemical Institute imeni L. Ya. Karpov)

PRESENTED: October 20, 1960, by A. N. Frumkin, Academician

SUBMITTED: October 13, 1960

S/020/62/143/003/026/029 B101/B144

18. 8360

AUTHORS:

Gil'man, V. A., and Kolotyrkin, Ya. M.

TITLE:

Mechanism of pitting corrosion of zirconium in halide

solutions .

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 143, no. 3, 1962, 640 - 642

TEXT: In the previous work (DAN, 137, no. 3, 642 (1960)) it as presumed theoretically that pitting corrosion of Zr in chloride solutions was a consequence of local depassivation of the metal surface by chlorine ions. This depassivation occurs when the Cl concentration reaches a ritical value, thus necessitating an induction period. This assumption was checked experimentally by measuring the time t_m (sec), which elapses after imposition of anodic polarization until the minimum ϕ_m occurs in

the curve φ versus t. Results at various current densities and electrolyte concentrations are (t_m, sec) :

Card 1/3

Mechanism of pitting corrosion...

s/020/62/143/003/026/029 B101/B144

	KBr			KCJ		
i,a/cm ²	0.01 N	0.1 N	1.0 N	0.01 N	0.1 N	1.0 N
5.10-6	1200	1080	-	1410	1290	-
5•10 ⁵	92.5	78.5	73.5	91	83.2	79•5
5•10 ⁻⁴	9.6	8.3	7.7	10.0	8.0	6.7

KCl + N	a ₂ S0 ₄	KCl + Na ₂ CO ₃		
8.	Ъ	С	ď	
2545 165 15.8	327' 38.9	174 16.5	250 26	

Legend: (a) 0.05 N[C1] + 0.05 N[S0 $_4^{2-}$] (b) 0.025 N[C1] + 0.075 N[S0 $_4^{2-}$] (c) 0.05 N[C1] + 0.05 N[C0 $_3^{2-}$] (d) 0.025 N[C1] + 0.075 N[C0 $_3^{2-}$]

With increasing halide concentration and decreasing current density the reproducibility of the data decreases. The mean deviation was in the case of $5 \cdot 10^{-4}$ a/om² and 0.01 N: 5 - 6%; of $5 \cdot 10^{-5} - 5 \cdot 10^{-6}$ a/cm² and 1.0 N: 18 - 23%. The zirconium specimens were treated with dilute HF. It is Card 2/3

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3 CIA-RDP86-00513R000515110005-3"

8/0020/64/155/005/1155/1158

ACCESSION NR: AP4034543

AUTHORS: Giliman, V. A.; Koloty*rkin, Ya. N. TITLE: The mechanism of dissolving zirconium in acid fluoride

solutions

SOURCE: AN SSSR. Doklady*, v. 155, no. 5, 1964, 1155-1158

TOPIC TAGS: zirconium, solution mechanism, dissolution kinetics, hydrogen evolution kinetics, zirconium oxidation, rate of solution

ABSTRACT: The kinetics of hydrogen evolution and the kinetics of zirconium dissolution were examined in this investigation to determine the mechanism by which zirconium is dissolved in acid fluoride solutions. Data was obtained from solutions of 0.1N $H_2SO_L + xHF$ where x is the concentration varied from 10-3 to 1N. Pyrex cells were used for $\int HF \int \leq 10^{-2}N$ and teflon or polyethylene cells for $\int HF \int \geq 10^{-2}N$; equivalence of results was claimed. The zirconium electrode preparation was described previously by the authors (DAN, 137, 642 (1961); DAN, 143, 640 (1962)). Solutions were nitrogen purged although air does not affect rate. The solution rate was determined by anode current and by colorimetry using xylenol orange APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3

ACC NR AP6015293

SOURCE CODE: UR/0365/66/002/003/0360/0361

AUTHOR: Gil'man, V. A.; Kolotyrkin, Ya. M.

ORG: Physicochemical Scientific Research Institute im. L. Ya. Karpov (Nauchnoissledovatel'skiy fiziko-khimicheskiy institut)

TITLE: Pitting corrosion of zirconium in perchlorate solutions

SOURCE: Zashchita metallov, v. 2, no. 3, 1966, 360-361

TOPIC TAGS: corrosion, zirconium, perchlorate, chloride

ABSTRACT: A study of zirconium corrosion in 0.1 and 1.0 N NaClO4 and HClO4 and also 0.3 N LiC104 showed that under spontaneous dissolution and anodic polarization conditions, zirconium is in a passive state until a certain critical potential $\phi_{\rm cr}$ is reached, at which extensive pitting begins to take place. In this respect, the anodic behavior of Zr in perchlorate solutions is similar to that in chloride solutions, except for the fact that in the latter the critical pitting potential is more positive by almost one whole volt. The value of ϕ_{cr} in perchlorate solutions is determined by the C104" concentration, increasing by 100 mV for a tenfold decrease of the perchlorate concentration, and, as in the case of chlorides, is independent of the solution pH or the anodic current density. Thus, halide ions are not the only ones to cause the pitting corrosion of mirconium; ClO4" ions also have this capacity (although not

Card 1/2

UDC: 620,193,01

L 04777-67 EWT(m)/EWP(w)/EWP(t)/ETI/EWP(k) IJP(c) JD/WW/JG/WB

ACC NRI AP6025725

SOURCE CODE: UR/0365/66/002/004/0490/0492

AUTHOR: Gil'man, V. A.; Kolotyrkin, Ya. M.; Malkine, R. I.

ORG: Scientific Research Physicochemical Institute im. L. Ya. Karpov (Nauchno-issledovatel'skiy fiziko-khimicheskiy institut)

TITLE: Solution of zirconium in concentrated hydrochloric acid

SOURCE: Zashchita metallov, v. 2, no. 4, 1966, 490-492

TOPIC TAGS: zirconium, corrosion, corrosion rate, electrochemistry, solution kinetics, chloride, induction melting, metal melting

ABSTRACT: Studies of the corrosion and electrochemical behavior of zirconium under anodic pole zation conditions were continued using concentrated HCl, 11.5 N. In the passive region, at potentials more negative than +0.17 v, the rate of Zr solution to Zr is independent of potential and amounts to 0.2-1.10 mp/cm. The rate of solution of Zr pre-etched in HF corresponds to the stationary anodic current density at the given potential. In the case of Zr with atmospheric oxide films, the initial average rate of solution is an order higher than the anodic current through the system, but becomes somewhat lower and almost constant with time. The proposed mechanism for the solution of passive

Card 1/2

UDC: 620-193-h1:669-296

Zr comprises the electrochemical formation of an oxide on the metal surface with subsequent solution of the oxide. At potentials above 0.17 v the rate of solution and anodic current increase rapidly resulting in embrittlement and eventual disintegration of Zr electrodes produced by induction melting. Action on arc melted Zr containing 0.02% C is ten times slower. Tests under potentiostatic conditions were found to be more severe than the corrosion tests run at 100°C. The rate of solution of Zr in concentrated HCl is 2 orders higher than in dilute acid. Orig. art. has: 2 figures.

SUB CODE: 07, 13/ SUBM DATE: 03Apr66/ ORIG REF: 003/ OTH REF: 004

Card 2/2 plas

GIL'MAN, Ya.

enrichten in the second Mechanization of cleaning and painting of vertical cylindrical tanks.

Stroitel' 2 no.4-5:33 Apply '56. (MIRA 10:1) Stroitel' 2 no.4-5:33 Ap My '56.
(Tanks) (Fainting, Industrial)

GILLIMAN, Ya.

Apparatus for grindles dand. Struttel' no.6:11 Je '57. (NERA 10:9) (Chalk) (Crushing machinery)

GILMAN, Ya., insh.

Precast reinforced concrete grain dryers. Gor.i sel.stroi. no.8/9:33 Ag S 157. (MIRA 10:1 (Precest concrete construction) (Grain elevators)

(MIRA 10:12)

CHEREPANTSEV, G., insh.; GIL'MAN, Ya., insh.

Demonstration building of a dormitory. Stroitel' no.4:3-4 Ap '58. (MIRA 11:5)

(Rostov-on-Don--Student housing)

GEL'FREYKH, V., arkhitektor; KORABEL'NIKOV, A., arkhitektor; GOLUBOVSKIY, L., arkhitektor; GIL'MAN, Ya., inzh.

> Design of an apartment house with rolled reinforced concrete components executed by the Institute for the Design and Planning of Housing and Civil Construction in the City of Moscow. Zhil. stroi. no.4/5:38-42 '58. (MIRA 12:6) (Apartment houses)

(Architecture-Designs and plans)

まれた 5/194/61/000/005/027/078 ...20】、203

18 6260 1527 1521

.JTHOR .

ART COLOR

TITLE:

Electric simulation of frames with offset joints taking into consideration the elastic fixing-in of struts

PERIODICAL:

Referativnyy zhurnal. Avtomatika i radioelektronika. no. 5, 1961, 34, abstract 5 B246 (Tr. 1-y mezhvuz nauchno-tekhn. konferentsii po elektr. modelirovaniyu zadach stroit. mekhan., soprotivleniya materialov i teorii uprugosti, B.m. Novocherk. politekhn.

in-t, 1960, 120-129)

TEXT: The possibility is considered of using the electric analogue PMT(1 (EMMS 1) for designing flat frames with offset joints and taking into consideration the effect of yielding of the base on the stresses resulting in the frame. It is emphasized that use of electric analogues reduces by many times the process of determining

Card 1/2

25/5 \$/194/61/000/005/027/078 D201/D303

Electric simulation.

stresses and obviates complicated calculations. When designing the analogue it is easy to take into consideration the effect of yielding of the base on bending moments within the frame. The discrepancies between the analytical and analogue results of calculations is about 5% which shows the accuracy and effectiveness of the method discussed. 10 figures. 8 references. Abstracter's note: Complete translation.

Card 2/2

GIL'MAN, Ya.D.

Preparing the foundation of a large-panel building on sagging soil. Osn., fund. i mekh. grun. 3 no.3:25-26 '61. (MIRA 14:7)

(Soil stabilization) (Foundations)

LOMIZE, G.M., doktor tekhn.nauk, prof.; GIL'MAN, Ya.D., inzh.

Electric spark method of compacting soil. Gidr. stroi. 32 no.6:42 Je '62. (MIRA 15:6)

(Soil stabilization)

LCMIZE, G.M., prof., doktor takhn. nauk; GIL'MAN, Ya.D., inzh.

Compacting soils by electric discharges. Trudy Giprovodkhoza no.22:155-162 '63. (MIRA 17:8)

GIL'MAN, Ya.D.

buildings and structures on seaging son, the son for the forigonian. Gan., fund. 1 mekb.grup. Hor, 150 th.

1-1,

YEGORSHIN, V.P., prof.; GIL'MAN, Ye.A., red.; VOZHESHNSKIY, A.D., tekhn.red.

[Theoretical mechanics; test assignments 1-4 for correspondence students in Course 2 with engineering majors in agricultural colleges] Teoreticheskais mekhanika; kontrol'nye sadaniis 1-4 dlia studentov-saochnikov II kursa inshenernykh spetsial'nostei sel'skokhosiaistvennykh vusov. [Version 7] Variant 7. Sostavil V.P.Egorshin. Moskva, 1958. 9 p. (MIRA 12:3)

1. Vsesoyusnyy sel'skokhosysystvennyy institut saochnogo obrazovaniya. (Mechanics--Problems, exercises, etc.)

TEGORSHIN, V.P., prof.; GIL'MAN, Ye.A., red.; VOZNESENSKIY, A.D., tekhn.red.

[Theoretical mechanics; test assignments 1-4 for correspondence students in course 2 with engineering majors in agricultural colleges] Teoreticheskaia mekhanika; kontrol'nye zadaniia 1-4 dlia studentov-zaochnikov II kursa inzhenernykh spetsial'nostei sel'skokhoziaiastvennykh vusov. [Version 3.] Variant 3. Sostavil V.P.Egorshin. Moskva, 1958. 10 p. (MIRA 12:3)

1. Vsssoyuznyy sel'skokhosysystvennyy institut zaochnogo obrasovaniya.
(Machanics--Problems, exercises, etc.)

ANDRIANOV, V.N., prof.; DRUZHININA, N.A., assistent; MISHARINA, Ye.A., kand.tekhn.nauk; BIKOHOV, L.V., dotsent; SHPRINK, B.E., prof., retsenzent; GLEBOVICH, A.A., kand.tekhn.nauk; GIL'MAN, Ye.A., red.; VOZNESHNSKIY, A.D., tekhn.red.

[Electric machines; instructions and assignments for students specializing in the electrification of agriculture] Elektricheskie mashiny; metodicheskie ukazaniia i kontrol'nye zadaniia dlia studentov spetsial'nosti "elektrifikatsiia sel'skokhoziastvennogo proizvodstva." Pod red. V.N.Andrienova i A.A.Glebovicha. Moskva, Mosk. in-t mekhanizatsii i elektrifikatsii sel'.khoz., 1958. 56 p. (MIRA 12:2)

(Electric machinery)

ZHURAVEL', I.V., dotsent; FLEKSER, Ya.N., doktor tekhn.nauk, red.; GIL'MAN, Ya.A., red.; VOZNESENSKIY, A.D., tekhn.red.

[Hydraulics; control lessons for correspondence students in engineering faculties majoring in irrigation and drainage] Gidravlika; kontrol'nye saniatiia dlia studentov-saochnikov gidromeliorativnoi spetsial'nosti inshenernogo fakul'teta.

Balashikha, 1959. 20 p. (MIRA 14:12)

1. Balashikha, Vsesoyusnyy sel'skokhozyaystvennyy institut saochnogo obrazovaniya. (Hydraulics)

TOMILIN, A.G., prof.; GIL'MAN, Ye.A., red.

[System of the animal world (characteristics of basic groups)]
Sistema zhivotnogo mira (kharakteristika osnovnykh grupp); uchebnos posobie dlia studentov zootekhnicheskogo i agronomicheskogo fakul'tetov. Moskva, Vses.sel'khoz.in-t zaochnogo obrazovaniia, 1962. 34 p. (MIRA 16:2)

(Zoology)

GIL'MANOV, G. R.

Qualified personnel for automatically controlled machinery. Prof.-tekh. obr. 20 no.4:28-29 Ap '63. (MIRA 16:5)

1. Direktor uehebno-kursovogo kombinata neftepromyąlovogo upravleniya "Oktyabr'skneft", Bashkirskaya ASSR.

(Petroleum workers--Education and training)

ANISINOV, Nikolay Geracimovich; Gil MONOV, Gilemdar Rizyanovich; STRATIYEV, Valentin lyanovich; GST MHE TRAYA, G.A., red.

[Frequency-type remote control system for oil fields] Chastotnaia sistema telemekhanizatsii neftepromyslov. Ufa, Bashkirskoe knizhnoe izd-vo, 1962. 83 p. (MIRA 17:7)

GIL'MANOV, G.R.; YURCHENKO, V.I.; SANDIKOV, A.V.

Determining the pressure on the intake of an electric centrifugal sinking pump by means of a frequency transducer. Nefteprom. delo no.9:26-29 165. (MIRA 18:10)

 Nauchno-issledovatel'skaya laboratoriya no avtomatike i telemekhanike neftepromyslovogo upravleniya "Oktyabr'skneft".

GIL'MANOV, G.R.

Means for reducing oil and gas losses in fields of the Petroleum Production Administration of the Association of the October Petroleum Industry. Nefteprom.delc no.10: 27-31 *65. (MHA 19:1)

1. Neftepromyslovoye upravleniye "Ortyabriskneft".

VALIULLIN, A.V.; (ILL'MANOV, I.G.; KHASANOV, Kh.Kh.; KOROL'CHUKA, V.M., red.; LODVIKOVA, A.S., red. izd-va; NABIULLINA, R.S., tekhn. red.

[Fruit culture of the Tatar A.S.S.R.] Sadovodstvo Tatarskoi ASSSR. Kazan', Tatarskoe knizhnoe izd-vo, 1960. 279 p. (MIRA 14:9)

(Tatar A.S.S.R.—Fruit culture)

YAKOVLEVA, V.I.; KRETOVICH, V.L.; GIL'MANOV, M.K.

Localization of glutamate dehydrogenase in corn roots. Biokhimiia 29 no.3:463-469 My-Ja '64. (MIRA 18:4)

1. Institut biokhimii imeni Bakha AN SSSR, Moskva.

TAKOVLEVA, V.I., KRETOVICH, V.L., GIL'MANOV, M.K.

Glutamic dehydrogenase of corn roots. Biokhimia 29 no.5: 896-904 Jl-Ag '64. (MIRA 18:11)

1. Institut biokhimii imeni Bakha AN SSSR, Moskva.

GIL'MANOVA, G.A.

Surplus callus formation following femoral fracture. Med.zhur. Uzb. no.11:68-70 N ¹60. (MIRA 14:5)

l. Iz Uzbekskogo nauchno-issledovatel skogo instituta travmatologli i ortopedii (direktor - A.Sh.Shakirov).

(FEMUR--FRACTURE)

GILTERTANT, G. RE., BOTRO, V.S., LAPSHEMA, G. C.

"The importance of gamesidee in the maintenance of a focus of tickborne encephalitie." Page 67

Desymbole soveshchamilye po parazitlogicheckin i roblemam i prirodnochagovym boleznyam. 22-29 Oktyabrya 1959 g. (Tenth Conference on Parasitological Problems and Dineases with Natural Poci 22-20 October 1959), Moscow-Leningrad, 1959, Academy of Medical Ociences USSR and Academy of Sciences USSR, D. 1 254pp.

SINJUNIOVA, S. KH., HOYEG, V. A., STITATOV, K. D., LAPTHIT, C. M., SWEALDULLI, YU. SH.

"The study of the natural fooi of tickborne one phriltic in the TASSR". Face 60

Designative sovemberive policipation in the conference is produced spovym belezivem. 25-29 Crtystrys 1959 g. (Tenth Conference in Foresithelical Froblems and Diseases with Natural Foci 22-29 Cotober 1959), Mascaw-Leningrad, 1969, Academy of Medical Sciences USOR and Academy of Sciences USOR, No. 1 254mm.

GIATERIOVA, G. Eh.; BOTKO, V.A.; LAFSERTA, G. .

insticipation of Camasidae mitos in the first of the bosse encophalitic virus in the natural first of the laws A.B.S.R. Red. paras. I paras. bel. 33 na.w:16-bandless 104 (1.50)

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GUMEROVA, M.Kh.; ARISTCVA, T.V.; GIL'MANOVA, R.G.; L'VOV, F.V.; BUKCHANTAYEVA, M.S.; MUKHAMETSHINA, M.A.; GATHULLINA, H.M.; KHRAMOVA, H.P.; KOHRANOVA, I.N., red.; LABUDIN, N.T., red.; IEROGIMOVA, Z.A., tekhn.red.

[Forty years of the Tatar A.S.S.R.; statistical collection]
Tatarakaia ASSR za 40 let; statisticheskii sbornik. Kazan;
Tatarakoe knishnoe izd-vo, 1960. 171 p. (MIRA 14:3)

1. Tatar A.S.S.R. Statisticheskoyo upravleniye. 2. Nachal'nik Statisticheskogo upravleniya Tatarskoy ASSR (for Kobranova). (Tatar A.S.S.R....Statistics)

Mitorial Basti E.M. Hostalov (Herp, Mi.) Profuser, A.A. Trufmore, (Resp. Mi.) Profuser, I. Th. Hoyse (Popty Brit.) Ed.) Profuser, G.S. Tesferi Gasally, Profuser, A. Te., Johnson, Anderson, A. Te., Johnson, Anderson, M. Haddenf, Profuser, B.M. Indirect., Frofuser, A.M. Gelger, Profuser, M.M. Roders, Profuser, B.M. Dollery, (Herp, Sections), M.M. Turnianson (Herp, Sections), M.M. I. M. Edgmillin, Turnianson fruct, 779. 22, ibinichentiys manid (frrunsetions of the Chenden sad Technological Invitetes immi B.M. Elvy, Kasas, Fg. 22, Chentell Stimmes) Emms, 1996. 17) p. Errais ally inseried. 300 upples printed. COVIDAGE: The exilaction contains reports by faculty sembers of the sponsoring include and also commencing the fitting and also commencing the fitting and first sentwenty of the fitting of frethered Address Alebest Milhopforted Familyev's scientific sentiation followed and good of the Pamilty. A retire of Teallyev's scientific sentiation in that of the fitting with a description of the particular sentiation where a sentiation of the interior PARCOR: This beek is intended for infantrial chanists, technologists, ectentists, beckening and recently action of the space of applied chanistry. ឌ ផ្ទ ğ Ħ 15. Terpin, H.E., and B.Y. Kollor. Density and Viscosity of the System Montan-Pleas. 16. Turpia, M.L., and B.A. Trifonov. Physicochamical Properties of the System Bioman-Value 13. Enmotest-Patisor, L.I., and E.B. Kranyr. Adacrytion of Hitrogen Codden 12. Departure, 1.5., and K.E. Nochalov. The Conversion of Nethans With Organ In an Electric Are Recharge 10. [11] research 9.0, and 71. M. forgin, The latitation of Corpses [Notice than the foreign than the foreign of land in a Marcary-Drop Electron (Frainciany report) 11. iiss'ell', 8.7., and Students 2.0. Senins and 1.0. Filtensors. The Presidility of Directly Determining Sodium in the Presence of Trengl Ress. 6102/2016 Keem, Brindle-tethnologisheelly inelibrations &R. Elvern 13. Aleksemfrova, To. I. Analysis of Tanalag Bathe Transactions of the Chemical (Cost.) 9/1

6.6.

GICMANSHIN,

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s/020/60/132/01/35/064 B011/B126

5.2400(A AUTHOR5:

Gil'manshin, G. G. Mochalov, K. N.,

TITLE:

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The Polarographic Behavior of Sodium-, Potassium-, and Lithium

Boron Hydrides

Doklady Akademii nauk SSSR, 1960, Vol. 132, No. 1, pp. 134-137 PERIODICAL:

TEXT: The views on the theme in the title are directly contradictory (Refs. 6,7) in the few (2) relevant works. In their experiments the authors used commercial ($\sim 80\%$) and purified (98%) boron hydrides. They used the micropolarograph of Heyrovsky, model M-102 with a dropping mercury electrode. For NaBH4 in NaOH they have found a single wave, namely that of the ion PH4. Its nature was determined by further experiments (Fig. 1, Table 1). The position and character of these waves remain practically unchanged through variations in the concentration of boron hydride and through changes in the composition of the background. This result disproves the data of R. L. Pecsok (Ref. 6). The authors studied the dependence of the height of the boron hydride wave on the concentration of BH_{$\frac{1}{4}$} ions. The dependence is linear between 10⁻³ and 10⁻¹ moles/1. The limiting current here is no complete diffusion current. The metallic boron hydrides

Card 1/3

The Polarographic Behavior of Sodium-, Potassium-, and Lithium Boron Hydrides

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decompose relatively quickly in aqueous, especially in acid solutions, so that the polarographing is made very difficult. Therefore, the solutions used were prepared with the use of the relevant alkalis and alkaline borate buffer mixtures. From this it was established that, for the same concentration, the wave height is highly dependent on the pH in the solution. With a pH above 12.5 the boron hydrides are relatively stable, but the wave was practically missing altogether. Thus, it follows that in reality the wave does not belong to the BH4 ion, but to one of its hydrolysis products. These occur in several stages in one of which diborane is given off under certain conditions. However, diborane can react with alkalis and form the so-called hypoborates (see scheme). Gaseous diborane was passed through concentrated KOH-, HaOH-, and LiOH solutions when cooled. The resulting hypoborate solutions showed the same wave with $E_1/2$ = = -0.6 v. The dilution of these solutions led to a proportional decrease in wave height. When the solution is left standing, the height of the "hypoborate" wave, exactly as the "boron hydride" wave, decreases according to an equation of the first order (Ref. 8). When the solutions are boiled and strongly acidified, the wave disappears after the destruction of the hypoborates. Thus, the "boron hydride" wave is basically a "hypoborate" wave. It is difficult to say to which of the 3 hypoborates the wave belongs. However, it cannot belong to the BH (OH)

Card 2/3

The Polarographic Behavior of Sodium-, Potassium-, and Lithium Boron Hydrides

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ion. It is more likely that the BH(OH); ion is responsible for the wave. The electrodic reaction which the said wave causes can obviously not (contrary to Pecsok) be brought about by oxidation of the BH, ions, but must be due to the oxidation of the hypoborate ions (see scheme). Of the two schemes set out, the second is more likely. The following are mentioned: D. Il'kovič, A. F. Zhigach, V. I. Mikheyeva, V. Yu. Surs, Kh. V. Shifrin, A. A. Bogonostsev, O. I. Rusetskiy, and T. N. Dymova. There are i rigure, 1 table, and 14 references,

ASSOCIATION: Institut obshchey i neorganicheskoy khimii Akademii nauk SSSR (Institute of General and Inorganic Chemistry of the Academy of Sciences, USSR)

PRESENTED: December 26, 1959, by I. I. Chernyayev, Academician

SUBMITTED: December 15, 1959

Card 3/3

37638 \$/076/62/036/005/013/013 B101/B110

11.1740

AUTHORS:

Mochalov, K. N., and Gil'manshin, G. G.

TITLE:

Polarographic study of alkali-metal boron hydrides

PERIODICAL: Zhurnal fizicheskoy khimii, v. 36, no. 5, 1962, 1089-1094

TEXT: With a view to elucidating the processes that occur in the hydrolysis of NaBH₄, KBH₄, LiBH₄, and C₅BH₄ solutions of these boron hydrides were examined polarographically in aqueous solutions by using a recording polarograph (type 7-77-46, "orion", Hungary), a mercury dropping electrode, and a calomel reference electrode. The boron hydrides were prevented from decomposing by being dissolved respectively in 0.2 M NaOH, KOH, and LiOH. Investigation of the polarization within the range +0.2 to -2.0 v at room temperature showed that, unlike what had been found by R. L. Pecsok (see below), the three boron hydrides gave rise to the same wave, namely $\mathbb{E}_{1/2} = -0.65$ v. Impurities (e.g., sodium alcoholates) did not affect $\mathbb{E}_{1/2}$. As a result of hydrolysis of the boron hydride, the wave amplitude decreased with time. This process can be accelerated by Card 1/3

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Polarographic study of alkali- ...

acidification, heating, or catalysis. Different backgrounds did not affect the wave. The wave $E_{1/2} = +0.105 - 0.013$ pH found by Pecsok is attributed to the anodic dissolution of Hg in an alkaline medium. Results: (a) Change in pH and temperature (15-35°C) do not affect the wave potential. The wave amplitude of $NaBH_4$ and KBH_4 in the range of 1.10^{-3} to 1.10^{-4} moles/l is a linear function of the concentration of boron hydride. (b) The wave amplitude decreases with increasing pH. At pH >12.5 - i.e., if no hydrolysis takes place at all - no further waves will appear. Polarographic analysis of CaH2 and B2H6 showed no wave with the first compound, but $E_{1/2}$ = -0.65 v when B_2H_6 was bubbled through NaOH or KOH. From this it is concluded that the wave is due to the resulting hypoborates. Polarographic results obtained from stepwise hydrolyzed LiBH, and from NaBH(OCH,) ; indicate that the wave is not produced by the BH ion but by the BH(OH) ion. Analysis of the polarographic kinetic curves for NaBH and KBH confirmed that the hydrolysis of these compounds V followed the theory of the kinetics of consecutive processes. There are Card 2/3

s/076/62/036/005/013/013 B101/B110

Polarographic study of alkali- ...

4 figures and 2 tables. The most important English-language reference is: R. L. Pecsok, J. Amer. Chem. Soc., 75, 2062, 1953.

ASSCCIATION: Kazanskiy khimiko-tekhnologicheskiy institut im. S. M.

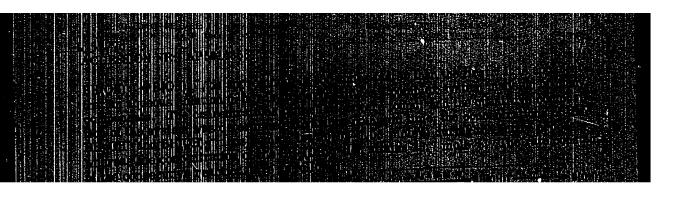
Kirova (Kazan' Institute of Chemical Technology imeni S. M.

Kirov)

SUBMITTED: August 19, 1961

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SATTAROV, M.M.; GIL*MANTHIN, I.G.

helling hassis of a human

Selection of wells for the carrying out of water-exclusion operations. Izv. vys. ucheb. zav.; neft¹ i gaz 6 no.7:43-47 163. (MIRA 17:8)

1. Ufimskiy neftyanov institut i neftepromyslovoye upravleniye "Arlanneft".

IMANAYEV, N.G.; GORGENER, B.Ya.; KRAVO LLET,: FLAURELL W. T.A.;

MARKOV, V.F.; SATTAROV, M.M.; GOLDVILL W., L.W.; GILLET, E.B.;

BORELYUK, V.F.; ROMANYOK, F.I.

Comments on the article by M.J. Contact v "Energy of reservoir waters". Neft, khoz., No.11, 19ch. Meft, ther. At m. 8:19-57 Ag 163.

Present status of and prespects for the coretraction of steel tanks in the U.S.B.R. Ibid.:58-45

1. Hefterromyslovoye ucravicnive Tuyman refit (for in 1917, Gombiner). 2. Heimskip refit by man man-finaled vate. skip instit ' (for Kravchenko, Black Lim). 3. Hefterromyslovove upravicnive Chernomorneft! (for Karkov). 4. Hefterromyslovove upravicnive Arlameft! (for Satterov, Girlmansida). 5. Germanstvennyv institut po provektirovaniva i issledivareljskim rabotam neftedobyvayushchey (romysmiennost) voorbehavk, rayonov strany (for Ashirov). 6. Vsecovannyv pefteramovyv nabelaosisslovovateljskiv institut (for behalivar, removia).

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3 APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3"

GILMEANU, Ion

After fourteen days the sections are reporting. Const Bue 16 no.732:2 18 Ja*64.

TROYANKIN, Yu.V., kand. tekhn. nauk; GIMMEL'FARB, M.L., dots., red.

[Methods for the design of a copper-melting reverberatory furnace] Metodika rascheta redeplavil'noi otrazhatel'noi pechi. Pod red. M.L.Gimmel'farba. Moskva, Mosk. energ. in-t, 1963. 30 p. (MIRA 17:4)

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3 APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3"

GORDON, M.K.; GIL'MOVSKAYA, M.I.

Clinical aspects of an atypical course in Addison-Biermer disease. Zdrav. Bel. 7 no. 4:43-45 Ap '61. (MIRA 14:4)

l. Iz terapevticheskogo otdeleniya (M.K. Gordon) oblbol'nitsy v g. Vileyka (glavnyy vrach A.S. Romashko). Nauchnyy rukovoditali raboty - professor G.Kh. Davgyallo. (ANEMIA)

GILLO, Georgiy Genrikhovich; MIL'CHIE, A.E., redsktor; VOLYNTSEVA, V.A., tekhnicheskiy redsktor.

[Laboratory work in the technology of typesetting by hand and machine] Laboratornye raboty po tekhnologii ruchnogo i mashin-nogo nabora. Moskva, Gos. izd-vo "Iskusstvo," 1954. 110 p.

(Typesetting) (MLRA 7:12)

[Continuity of production in the Eve Sokolova typesetting plant] Potochnost' proizvodstva v nahornom tsekhe tipografii imeni Mvg. Bokolovci. Moskva, Gos.iza-vc "lakuastvo," 1957. S2 o. (Typesetting) ACC NRI AT7000188

SOURCE CODE: UR/0000/64/000/000/0162/0170

AUTHOR: Volodarskiy, R. F.; Gilod, D. A.; Demidova, M. A.

ORG: none

TITLE: Sketch map of the present-day surface of the folded basement of the Ciscaucasus from geophysical data

SOURCE: Moscow. Universitet. Kafedra geofizicheskikh metodov issledovaniya zemnoy kory. Geofizicheskiye issledovaniya (Geophysical research), no. 1. Moscow, Izd-vo Mosk. univ., 1964, 162-170

TOPIC TAGS: 'earth crust, gravity survey, magnetic survey/Russian platform,

Ciscaucasus

ABSTRACT: Comprehensive analysis of geologic, geophysical, and borehole materials, as well as analysis of gravity and magnetic maps recomputed for different levels of the upper half space, have resulted in a tectonic regionalization of the Ciscaucasus and the solution of problems dealing with the geologic structure of the area. The article contains maps of the tectonic zoning of the folded basement of the Ciscaucasus and the southern regions of the Russian platform and surface of the Paleozoic basement of the Ciscaucasus are given. Orig. art. has: 2 figures.

SUB CODE: 08/ SUBM DATE: 05Nov64/ ORIG REF: 013/

Cord 1/1

Investigation of a reduction in well-priton permeability based on the pressure buildup curves. Nefterrom. Solo no. 10.48 to3.

1. TSekh nauchno-isaledovateliskikh i protwodstverrykh robot mai tepromyalovogo epravleniya "Ordobonikidaenefita.

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3 CIA-RDP86-00513R00051510005-3 CIA-RDP86-00513R00051510005-3 CIA-RDP86-00513R00051510005-3 CIA-RDP86-00513R00051510005-3 CIA-RDP86-005100005-3 CIA-RDP86-005100005-3 CIA-RDP86-0051000005-3 CIA-RDP86-0051000000005-3 CIA-RDP86-0051

Determining the optimal disbursement of a demalsifier from the data of an investigation of compressor wells. Nefteprom. delo no.10:37-40 164. (MIRA 17:12)

1. Heftepromyslovoye upravleniye "Ordzhonikidzeneft".

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3 CIA-RDP86-00513R000515110005-3 CIA-RDP86-00513R000515110005-3

New method for the automatic control of the level of the oil-water interface in the Lobkovo horizontal sedimentation tanks. Nefteprom. (MIRA 18:3)

1. TSekh nauchno-issledovatel*skikh i projzvodstvennykh rabet neftepromyslovogo upravleniya "Ordzhonikidzeneft*".

Flows

New plows PKB-56P and PKB-2-54 for brush and swamp ground. Sel'khozmashina No. 9, 1952

"APPROVED FOR RELEASE: Thursday, September 26, 2002
APPROVED FOR RELEASE: Thursday, September 26, 2002
CIA-RDP86-00513R000515110005-3

KALYUZHNYY, G.D., GIL'SHTRYN, P.M.

POB-3-45 plow for reclaimed swamps. Sel'khosmashina no.11:12-13 M 155. (Plows) (MLRA 9:1)

GIL'SHTEYN, P.M., inshener.; STARODINSKIY, D.Z., inshener.

New brush and bog plows. Sel'khosmashina no.4:5-6 Ap '57. (MLRA 10:4)
(Plows)

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3"

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CIA-RDP86-00513R000515110005-3"

CIA-RDP86-00513R000515110005-3"

CIA-RDP86-00513R000515110005-3"

Automatic equipment for mounting machines on tractors. Trakt. i sel'khosmash. no.11:13-16 E *58. (MIRA 11:11)

(Agricultural machinery)

GIL'SHTHYN, P.M., insh.; STARODINSKIY, D.Z., insh.

The PBM-2-60 mounted brush-breaker and bog plow. Trakt. i sel-khosmash. no.1:38-39 Ja 159. (Plows)

411 Showard, P.P.; STLRODITSKIY, D.Z. The rest of the second

Brush-propies plow. Trott.1 selfchormsh. no.7:33-3h Jl tso.

(BIR. 17:11)

1. Spetsiclinova konstruktovekove bymno nevode im si Omymbrisky su manolymisil.

(Plows)

Mounted scarifier for cultivating soil before deep plowing. Trakt.i sel'khozmash. no.10:30 0 '59.

(MIRA 13:2)

1. Spetsial'noye konstruktorskoye byuro zavoda im. Oktyabr'skoy revolyutsii.

(Agricultural machinery)

GIL'SHTEYN, P.M., [Hil'shtein, P.M.]; STARODINSKIY, D.A. [Starodyns'kyi, D.Z.], insh.

Mounted two-bottom brush-breaker plow. Mekh.sil'.hosp. 10 no.12:24-25 D 159. (NIRA 13:3)

GIL'SHTEYN, P.M., STARODINSKIY, D.Z.

Mounted cultivator and scarifier for stony soils. Trakt. i sel'khozmash. 30 no.8:37 Ag '60. (MIRA 13:8)

"APPROVED FOR RELEASE: Thursday, September 26, 2002
APPROVED FOR RELEASE: Thursday, September 26, 2002
GIL'SHTEYN, P.M., insh.; BLOSHTEYH, B.V., insh.

Hounted mulch-culture cultivator with subsurface sweeps. Trakt. i sel'khozmash. 30 no.11:32-33 N '60. (MIRA 13:12) (Cultivators)

"APPROVED FOR RELEASE: Thursday, September 26, 2002
APPROVED FOR RELEASE: Thursday, September 26, 2002
GIL'SHTEYN, P.M. [Hil'shtein, P.M.], inzh.; BLOSHTEYN, Ye.V. [Bloshtein,

KPL-2-100 cultivator with subsurface sweeps. Mekh. sil!. hosp. 12 no. 5:22-23 My 161. (MIRA 14:5)

l. Odesskiy zavod im. Oktyabr'skoy revolyutsii. (Cultivators)

"APPROVED FOR RELEASE: Thursday, September 26, 2002 APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3 CIA-RDP86-00513R000515110005-3"

GIL'SHTEYN, P.M.; STARODINSKIY, D.Z.

Increase in the traction indices of a wheel-type tractor operating with a mounted plow. Trakt.i sel'khozmash. 32 no.9:16=18 S '62.

(MIRA 15:12) 1. Spetsial noye konstruktorskoye hyuro zavoda imeni Oktyabr'skoy revolyutsii.
(Tractors)

(Plowing)

GIL'SHTKYN, F.M.; STARODINSKIY, D.Z.

Single-frame plows for brush and swamp lands. Trakt, i sel'khozmash. 31 [i.e.32] no.11:33-34 N '62. (MIRA 15:12)

1. Spetsial'noys konstruktorskoys byuro savoda imeni Oktyabr'skoy Nevolyutsii.

(Plows)

GIL'SHTEYN, P.M.; STARODINSKIY, D.Z.; TSIMMERMAN, M.Z.;

DOGANOVSKIY, M.G., kand. sel'khoz. nauk, retsenzent;

BUD'KO, V.A., inzh., red.

[Tillage machines for special purposes; their design and calculation] Pochvoobrabatyvaiushchie mashiny spetsial:nogo naznacheniia; proektirovanie i raschet. Moskva, lzdvo "Mashinostroenie," 1964. 139 p. (MIRA 17:11)

1. Vedushchiy konstruktor Spetsial'nogo konstruktorskogo byuro zavoda sel'skokhozyaystvennogo mashinostroyeniya im. Oktyabr'skoy revolyutsii (for Gil'shteyn, Starodinskiy, TSimmerman).

"Rocket Motors," M. Defense Publ. House, 1950

GILTCO, B.; DIACENKO, F.; E'ELIN, A.

Use of radioactive isotopes for determination of damage to machine parts. p. 3. TEHNICA NCUA. (Asociatia Stiintifica a Inginerilor si Tehnicienilor) Rucuresti. Vol. 2, No. 25, Novi 1955.

So. East European Accessions List Vol. 5, No. 9 September, 1956

NESHUMOV, B.; GIL TER, I.

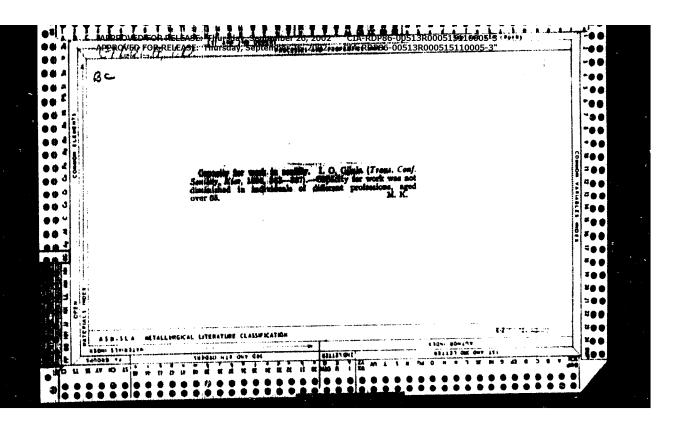
Turned wood toys. IUn.tekh. no.6:49-50 Je '57. (MIRA 10:7)
(Toys)

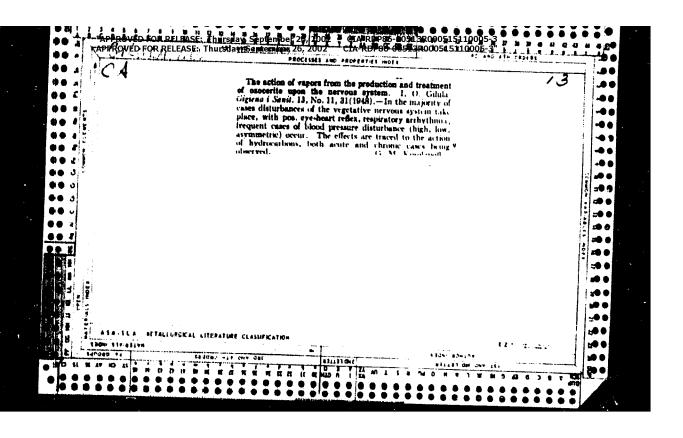
GILULA, M., kand. tekhn. nauk

Hequirements for dump trucks used in excavation operations.

Avt. transp. 38 no. 5:13-14 My 160. (MIRA 14:2)

(Dump trucks)





"APPROVED FOR RELEASE: Thursday, September 26, 2002
APPROVED FOR RELEASE: Thursday, September 26, 2002
GILOTA, I.O.; TSAPENKO, Y.L.

CIA-RDP86-00513R000515110005-3"

ALE INTERNATIONS OF THE PARTY O

Outaneous temperature in lesions of the cerebral cortex. Zhur.nevr.i paikh. 53 no.11:878-881 H '53. (MLRA 6:12)

1. Hafpdra nervnykh bolemney Kiyevskogo meditsinskogo stomatologicheskogo instituta.

(Brain-Disease) (Temperature, Animal and human)

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3 APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3"

GILULA, I.O., professor (Kiyev)

Penicillin for treating infectious diseases of the nervous system.

Vrach.delo no.4:431 Ap '57. (MLRA 10:7)

(PRNICILLIN) (MERVOUS SYSTEM--DISEASES)

GILULA, I.O., prof. (Kiyev); MOVIK, I.O., prof. (Kiyev); TSAPENKO, Ye.L., kand, med. nauk (Kiyev)

Higher nervous activity in patients with paradentosis. Probl. stom. 4:7-14 158. (MIRA 13:6)

"APPROVED FOR RELEASE: Thursday, September 26, 2002
APPROVED FOR RELEASE: Thursday, September 26, 2002
GILULA, I.O., prof. (Klyev)

CIA-RDP86-00513R000515110005-3

Bystem. Vrach. delo no.5:493-406 Ky 158 (MIRA 11:7)
(GENERATIVE ORGANS--DISEASES)
(NERVOUS SYSTEM--DISEASES)

GIDULA, I.O., prof. (Kiyev)

"Neuroses of visceral etiology", by O.R. Kirichins'kii. Reviewed by I.O. Gilula. Vrach.delo no.9:995-997 S'58 (MIRA 11:10) (NEUROSES) (KIRICHINS'KII.O.R.)

GILULA, I.O., prof.

Disability evaluation for persons with vascular diseases of the brain. Vrach.delo no.2:139-143 F '60. (MIRA 13:6)

1. Gorodskaya klinicheskaya bol'nitsa, Kiyev.
(DISABILITY EVALUATION) (BRAIN--DISEASES)

"APPROVED FOR RELEASE: Thursday, September 26, 2002
APPROVED FOR RELEASE: Thursday, September 26, 2002
GILULA, I.O., prof. (Kiyev) CIA-RDP86-00513R000515110005-3 CIA-RDP86-00513R000515110005-3"

"Problem of brain development and the effect of harmful factors on it." Reviewed by I.O.Gilula. Vrach. delo no.12:151 D '61.

(MI:W 15:1)

(BRAIN)

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3 APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515110005-3"

GILULA, Isaak Osipovich, prof.; KIRYUCHINSKIY, A.R., red.; BYKOV, N.M., tekhn. red.

[Neuropathology in gynecology; applicable in disorders of the ovarian-menstrual cycle, diseases of the uterus and ovaries during the periods of sexual maturation and the climacteric]
Nervnaia patologiia v ginekologii; pri narusheniiakh ovarial'no-menstrual'nogo tsikla, zabolevaniiakh matki i iaichnikov v period polovogo sosrevaniia i klimaksa. Kiev, Gos. med. izd-vo USSA, 1961. 61 p. (MIRA 15:3)

(GENERATIVE ORGANS, FEMALE--DISEASES)
(CLIMACTERIC) (PUEERTY)

GILULA, I.O.; STAL-NENKO, Ye.2. (Elger)

Characteristics of the narvous system in newtorn infants under normal and pathological conditions. ner. a vr. 1 psikh. 63 no.7:1012-1017 *63. (MiRA 17:7)

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